|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| aluminum | Al | helium | He | radon | Rn |
| antimony | Sb | hydrogen | H | rubidium | Rb |
| argon | Ar | indium | In | scandium | Sc |
| arsenic | As | iodine | I | selenium | Se |
| barium | Ba | iron | Fe | silicon | Si |
| beryllium | Be | krypton | Kr | silver | Ag |
| bismuth | Bi | lead | Pb | sodium | Na |
| boron | B | lithium | Li | strontium | Sr |
| bromine | Br | magnesium | Mg | sulfur | S |
| cadmium | Cd | manganese | Mn | tellerium | Te |
| calcium | Ca | mercury | Hg | thallium | Tl |
| carbon | C | neon | Ne | tin | Sn |
| cesium | Cs | nickel | Ni | titanium | Ti |
| chlorine | Cl | nitrogen | N | tungsten | W |
| chromium | Cr | oxygen | O | uranium | U |
| cobalt | Co | palladium | Pd | vanadium | V |
| copper | Cu | phosphorus | P | xenon | Xe |
| fluorine | F | platinum | Pt | zinc | Zn |
| gallium | Ga | plutonium | Pu | zirconium | Zr |
| germanium | Ge | potassium | K |  |  |
| gold | Au | radium | Ra |  |  |
| **In addition to the above, you should know which elements are** | | | | |  |
| **diatomics. \*\*\*These elements exist in pairs, never as single** | | | | |  |
| **elments. They are: I2, Br2, Cl2, F2, O2, N2, H2.** | | | |  |  |